

Answer Key To Word Equations

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Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead. $Zn + Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$ Single Replacement 2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas. $2AlBr_3 + 3Cl_2 \rightarrow 2AlCl_3 + 3Br_2$ Single Replacement 3.

Worksheet: Word Equations KEY Name Substitute symbols and formulas for words, then balance each equation. 4. sodium chloride + lead (II) nitrate lead (II) chloride + sodium nitrate

WS6-2 Word Equations.doc 3 Answers to Worksheet 6.1 Writing and Balancing Equations 1. Write the chemical equations and balance each of the following word equations. a) Aluminum metal reacts with iron (II) oxide powder to produce aluminum oxide solid and iron metal. $2Al(s) + 3FeO(s) \rightarrow Al_2O_3(s) + 3Fe(s)$

Math 8: One-Step Equations Word Problems (Writing Algebraic Expressions) Write an expression for each word phrase. 1. a number x divided by 7 2. the sum of 10 and a number n _____ 3. a number b decreased by 14 4. the product of a number x and 8

Writing and Balancing Equations Worksheet Word Equations – Answer Key 1) Zinc and lead (II) nitrate react to form zinc nitrate and lead. $Zn + Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$ _____ 2) Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas.

Balancing Chemical Equations Word Problems Name Substitute symbols and formulas for words, then balance each equation. Be sure to write the chemical equations correctly BEFORE adding the coefficients to balance them. 1 sodium chloride + lead (II) nitrate lead (II) chloride + sodium nitrate $2NaCl + Pb(NO_3)_2 \rightarrow PbCl_2 + 2NaNO_3$

Word Equations Worksheet W 327 Everett Community College Student Support Services Program Write the equations for the following chemical reactions: 1) When dissolved barium chloride reacts with dissolved potassium sulfate in water, barium sulfate precipitate and aqueous potassium chloride are made.

Pairs Check: Systems of Equations Instructions: Partner 1 will do the problem in the left column while Partner 2 coaches. Then, Partner 2 will do the problem in the right column while partner 1 coaches. When you reach a STOP POINT, raise your hand to get a teachers signature.

Complete the equations. 0.5 4 5 0.05 0.05 3 5 0.5 2 Draw a number line from 0 to 2. Then draw and label points at 2 and 0.2. Use the number line to explain why 2 is 10 times the value of 0.2. Possible answer: The number 2 is 10 times the value of 0.2 because 2 is 10 times as far from 0 as the distance from 0.2 to 0. Answers will vary.

Title: Microsoft Word - worksheet - writing chemical equations review.doc Author: Glenn McCabe Created Date: 11/21/2010 6:44:17 PM

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Worded Simultaneous Equations (new GCSE) New GCSE Maths (higher / foundation)